

# (K, Na)NbO<sub>3</sub>-based lead-free piezoceramics: Phase transition, sintering and property enhancement

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**Abstract:** Most widely used piezoelectric ceramics are based on Pb(Zr,Ti)O<sub>3</sub> (PZT) composition which has adverse environmental and health effects due to its high lead content. Environmental and safety concerns with respect to the utilization, recycling, and disposal of lead-based piezoelectric ceramics have induced a new surge in developing lead-free piezoelectric ceramics. Among all the lead-free ceramics, (K,Na)NbO<sub>3</sub> (KNN) has drawn increasing attention because of its well-balanced piezoelectric properties and better environmental compatibility. On basis of the author's work, this review summarizes the progress that has been made in recent years on development of KNN-based piezoelectric ceramics, including crystallographic structure and phase transition analysis, pressurized solid-state sintering as well as liquid-phase-assisted sintering process, and poling treatment for property enhancement. All in all, KNN is a promising lead-free system, but more research is still required both from academic and industrial interests.

**Key words:** electrical properties; ferroelectric properties; piezoelectric properties; perovskites; niobates; lead-free

## 1 Introduction

In the 1880s, Pierre and Jacques Curie discovered that some crystalline materials, when compressed, produce a voltage proportional to the applied pressure. This characteristic is called piezoelectricity [1]. Although piezoelectricity is found in several types of natural materials, most modern devices use polycrystalline ceramics. Piezoelectric ceramics, which play an important role as functional electronic materials, have

been widely used in various applications such as sensors, actuators transducers and so on. For decades, lead zirconate titanate (Pb(Zr<sub>1-x</sub>Ti<sub>x</sub>)O<sub>3</sub> or PZT) ceramics has been market-dominating due to its excellent properties. Generally, the commonly used compositions are along a morphotropic phase boundary (MPB) around  $x = 0.48$  separating tetragonal and rhombohedral phases, where dielectric, piezoelectric, and electromechanical coupling properties are optimized. The origin of the enhanced properties of PZT at the morphotropic phase boundary has been investigated in detail by many authors and is still a topic of intense interest. Several mechanisms have been invoked to explain the enhanced piezoelectric properties of PZT at MPB. However,

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there is still no consensus in the literature on this issue. A widely accepted perspective believes that the increased properties are due to enhanced polarizability arising from the equivalent energy state between tetragonal and rhombohedral phases, which underlies optimal domain rotation during the poling process.

However, the large amount of lead contained in PZT materials has drawn much attention during the past decade, due to the environmental concern as well as government regulations against hazardous substances. In 2006, the European Union has already adopted the well-known directives Waste Electrical and Electronic Equipment (WEEE) and Restriction of the use of certain Hazardous Substances in electrical and electronic equipment (RoHS), with the purpose of protecting human health as well as environment by exclusion or substitution of hazardous substances used in electrical and electronic devices [2,3]. Similar regulations are also planned or established in North America, Japan, Korea and China. Lead ranks foremost among the hazardous substances list due to its notoriety, with maximum allowed concentration of mere 0.1 wt%. However, PZT, which contains more than 60 wt% lead oxides, is still allowed to be used in piezoelectric devices all over the world up to now, since no alternative lead-free counterparts is available. Therefore, tremendous efforts have been devoted to the development of competitive lead-free counterparts, such as BaTiO<sub>3</sub>, Bi<sub>1/2</sub>Na<sub>1/2</sub>TiO<sub>3</sub>-based perovskites, bismuth layer-structured ferroelectrics (BLSF), and (K, Na)NbO<sub>3</sub> (KNN)-based perovskites, and so on. At the moment, great embarrassment has been encountered in search for one single qualified lead-free system substituting PZT, which is so versatile in properties by utilization of various doping elements that it can fit almost all purposes. Therefore, it is reasonable to penetrate deeply into these various lead-free systems, and take advantage of them in distinct circumstances.

Among all the lead-free candidates, alkali niobate ceramics based on KNN have drawn much attention since Saito *et al.* made the breakthrough in Li, Ta, and Sb-modified KNN ceramics with textured structure [4]. Despite pure KNN ceramics is normally considered with a piezoelectric constant  $d_{33}$  as low as around 80 pC/N, the value reported by Saito *et al.* reached amazingly up to 416 pC/N, comparable to that of soft PZT ceramics. At first, it is believed that dopants like Li, Ta and Sb could consist in the constitution of MPB in KNN-based ceramics, which mimics that in PZT

ceramics and eventually results in the property enhancement. Thus, all kinds of dopants are exploited with special emphasis on the improvement of piezoelectric constant  $d_{33}$ , which is usually measured using the convenient quasi-static method (Berlincourt-meter method) and taken as the small signal  $d_{33}$ . Generally, the most effective approach for property enhancement of KNN-based ceramics is to form solid solutions with other species, such as LiNbO<sub>3</sub> [5-12], LiTaO<sub>3</sub> [7,13-20], LiSbO<sub>3</sub> [21-23], BaTiO<sub>3</sub> [24,25], CaTiO<sub>3</sub> [26-30], (Bi<sub>0.5</sub>Na<sub>0.5</sub>)TiO<sub>3</sub> [31-35], or a combination of multiple additives [30,36-51], and typical  $d_{33}$  values obtained are among 200~300 pC/N. Later on, it is proposed that the underlying mechanism for property enhancement of KNN-based ceramics by means of above-mentioned manners is not classical MPB but PPT, which is the abbreviation of polymorphism phase transition and exists in many materials such as BaTiO<sub>3</sub> [15]. The key point of PPT theory claims that property enhancement in KNN-based ceramics is due to tetragonal-orthorhombic polymorphism phase transition point ( $T_{O-T}$ ) shifting downwards from around 200 °C to room temperature. However, MPB is actually more favorable over PPT since PPT has a problem of temperature instability [27,46]. As a result, the development of KNN-based ceramics is retarded for the time being, the situation of which has been encountered before by many great scientific innovations. Fortunately, several tentative ideas have already been brought out to further exploiting the potential of KNN-based materials as next generation lead-free piezoceramics [52-54].

Technically, it seems that the situation in KNN-based system is more complicated than that of traditional PZT series, which has invoked so many interesting scientific discussions. The authors have been working on KNN-based ceramics for almost 10 years, and would like to share these little tiny experiences with those who have already working on or will jump into this field. Certainly, there are already excellent reviews with regard to lead-free piezoceramics [55-60], but none of them is dedicated to KNN-based ceramics. The present review covers topics over a broad range, including crystallographic structure, sintering techniques and poling tricks. Nevertheless, up to now the perception of KNN-based materials is still not in-depth, thus divergence on these issues may be found in literature.

## 2 Polymorphism and Property Enhancement

KNN ceramics were investigated intensively several decades ago both in material properties and structures [61,62], as it was found as a piezoelectric material in as early as 1950s [63]. It is well known that the most interesting compositions are located with K/Na ratio approximately equal to 1:1 [63-65], and the KNN ceramics mentioned hereafter refer to these compositions if no specific explanations provided. KNN ceramics have an orthorhombic structure with space group  $Amm2$  at room temperature; however, it is a little complicated because it holds an orthorhombic structure while the perovskite type  $ABO_3$  subcell possesses monoclinic symmetry [61]. Furthermore, there is no standard JAPDS-ICDD files for exact  $K_{0.5}Na_{0.5}NbO_3$  even up to now. As a result, crystallographic indexing is inconsistent in literature, especially for researchers mainly focusing on materials properties. So what really happens to the structure of KNN-based ceramics is not clear enough. Furthermore, the enhanced piezoelectric properties in the modified KNN were firstly attributed to the formation of the MPB similar to that observed in the PZT system [5,11,36,45], which was later on pointed out to be a different effect namely polymorphic phase transition (PPT) [15,22]. In any case, phase structure investigations in KNN-based ceramics are of paramount importance for obtaining excellent properties.

Generally, pure KNN ceramics have the following polymorphisms, low temperature ( $<123^\circ\text{C}$ ) rhombohedral (R) phase, room temperature orthorhombic (O) phase, high temperature tetragonal (T,  $200\sim410^\circ\text{C}$ ) and cubic (C,  $>410^\circ\text{C}$ ) phases [66]. Also, these phase transition temperatures can be tuned by introduction of dopants, such as Li, Ta, Sb, etc. In order to elucidate the crystalline structure of KNN and investigate the influence of dopants on the phase evolution as well as properties enhancement,  $(1-x)(K_{0.5}Na_{0.5})NbO_3-xLiNbO_3$  ceramics were taken as an example in the following context to correlate material structures with piezoelectric properties [67]. Hence, a better understanding of phase structure and its relation with properties in KNN-based ceramics may be attained, which could possibly benefit researchers for properties enhancement.

The structure illustration for room temperature KNN is shown in Fig. 1. Although the structure of KNN at room temperature is orthorhombic, the perovskite type  $ABO_3$  subcell, as shown in Fig. 1a,

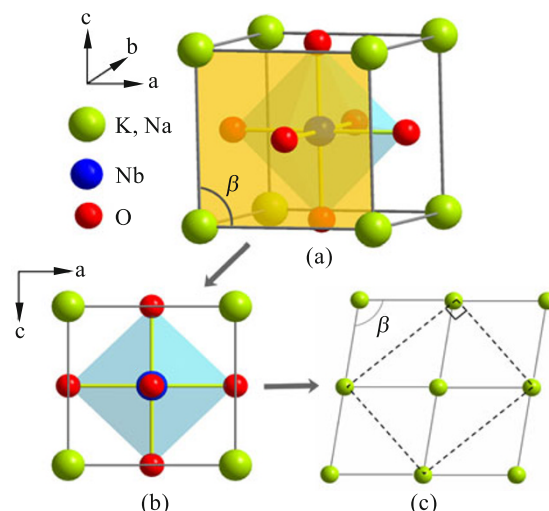


Fig. 1 The structure illustration for KNN of orthorhombic structure at room temperature (a) the perovskite type  $ABO_3$  subcell; (b) the projection of the subcell along  $b$  axis; (c) four adjacent subcell projections in an exaggerated manner to show the geometrical relationship, while omitting Nb and O atoms. (Reprinted with permission from Ref. [67]. © 2007, American Institute of Physics)

possesses monoclinic symmetry, with lattice parameters  $a_m = c_m > b_m$  while  $b_m$  axis is perpendicular to  $a_m c_m$  plane and angle  $\beta$  a little more than  $90^\circ$ . For the composition  $(K_{0.5}Na_{0.5})NbO_3$ , which is believed to lie along the orthorhombic-orthorhombic MPB and possess the best performance in  $KNbO_3$ - $NaNbO_3$  system, no superlattice reflections were found in previous XRD measurements [61], and thus no corner-linked O octahedra tilting should be taken into account [68]. Projection view of the subcell along  $b_m$  axis is shown in Fig. 1b. Since angle  $\beta$  is very close to  $90^\circ$ , it looks like that  $a_m$  axis is also perpendicular to  $c_m$  axis. Therefore, in order to show the geometrical relationship more clearly, we could exaggerate angle  $\beta$  much more than  $90^\circ$ , and draw the projection of four adjacent perovskite subcells together, but omit Nb and O atoms, as shown in Fig. 1c. As  $a_m$  axis equals  $c_m$  axis in length, the diagonals linked by dashed lines in Fig. 1c form a rectangle, which is the projection of unit cell of KNN along  $b_m$  axis. Hence it could be easily understood that the perovskite type subcell of KNN is monoclinic while its unit cell has orthorhombic symmetry at room temperature. Furthermore, when temperature increases above  $200^\circ\text{C}$ , a phase transition from orthorhombic to tetragonal occurs in KNN [69], which is similar to that of  $KNbO_3$ , with space group

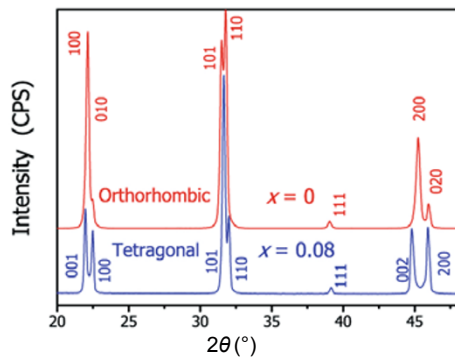


Fig. 2 The X-ray diffraction patterns of  $(1-x)(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3-x\text{LiNbO}_3$  ceramics

changing from  $\text{Amm}2$  to  $\text{P4mm}$  [70]. It is considered that dopants substitution, such as Li, in KNN would cause similar phase transition at room temperature. So in the tetragonal phase of Li-modified KNN, the perovskite type  $\text{ABO}_3$  subcell is the unit cell itself, with lattice parameters  $c_t > a_t = b_t$  and all three axes perpendicular with each other.

The above mentioned distinction between orthorhombic and tetragonal phases in KNN-based ceramics could be characterized by X-ray diffraction (XRD). The XRD patterns of  $(1-x)(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3-x\text{LiNbO}_3$  ceramics are shown in Fig. 2, and the crystallographic indexing is according to the perovskite type  $\text{ABO}_3$  subcell, both for orthorhombic structure when  $x = 0$  and tetragonal structure when  $x = 0.08$ , respectively. It is certain that the structure variation is due to the large distortion caused by doping Li element. The difference of peak profiles around  $45^\circ$  in X-ray diffraction pattern is usually used as an evidence for the phase transition in literature, though why these peaks would change in such a manner, or what really happens to the structures is hardly mentioned. The phase transition of  $(1-x)(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3-x\text{LiNbO}_3$  ceramics could be deduced from profiles of the  $N = 12$  ( $\{hkl\} = \{222\}$ ) and  $N = 16$  ( $\{hkl\} = \{400\}$ ) line groups in XRD patterns, as shown in Fig. 3. The splitting of the  $N = 12$  line group, as shown in Fig. 3a, indicates the existence of angle  $\beta$  in the monoclinic perovskite subcell. The two peaks around  $84^\circ$  in the XRD pattern when  $x = 0$  tend to merge into one single peak with increasing Li amount, responding to the decrease in angle  $\beta$ . On the other hand, the  $N = 16$  line group, as shown in Fig. 3b, shows the change of the three axes length in the subcell. It is obvious that in the orthorhombic phase when  $x=0$ , the intensity of the peak near  $100^\circ$  is larger than that of the one near  $103^\circ$ ,

because  $a_m = c_m > b_m$ ; while in the tetragonal phase the opposite situation occurs due to  $c_t > a_t = b_t$ . Therefore, when  $x = 0.06$ , the two contiguous peaks around  $84^\circ$  and the widely broadening peak around  $99^\circ$  obviously represent an intermediate state, showing the two phases co-existence condition. In addition, Raman spectroscopy is also an useful and convenient technique for phase identification of KNN-based piezoceramics [10,15,65,66,71-73].

As for property enhancement, shifting downward of tetragonal-orthorhombic transition ( $T_{\text{O-T}}$ ) point to room temperature by doping or forming solid solutions in KNN with other species do increase piezoelectric constant  $d_{33}$  to a level of 200~300 pC/N, while pure KNN ceramics suffering a relative low  $d_{33}$  around 80 pC/N. The fatal drawback is temperature instability [27, 46], which is absent in PZT systems due to existence of genuine MPB other than PPT. Apart from concerns about temperature instability of properties, it is noted that for identical or similar compositions of KNN-based ceramics, obtained properties could vary a lot among different research groups, which is possibly due to the exact location of  $T_{\text{O-T}}$ . Recently, it is reported

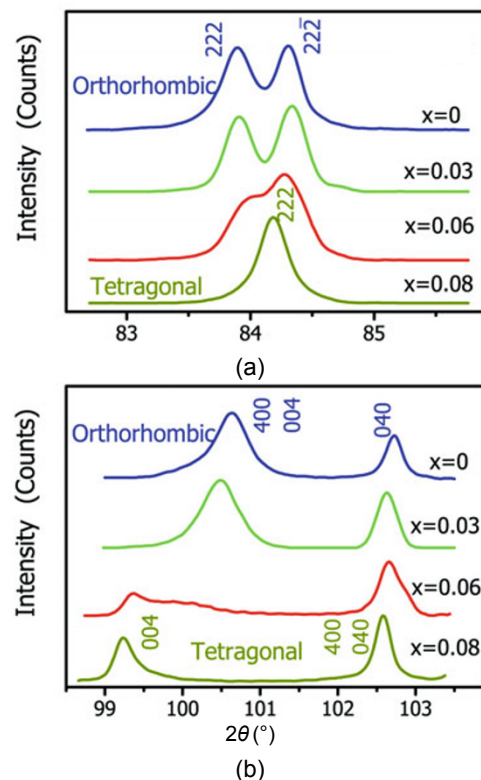


Fig. 3 The X-ray diffraction patterns of  $N = 12$  and  $N = 16$  groups in  $(1-x)(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3-x\text{LiNbO}_3$  ceramics



that the  $T_{O-T}$  could even be influenced as much as tens of degrees due to intergranular stress [73]. Besides, property enhancement in KNN-based ceramics appears to come a halt, which is actually the bottleneck effect of PPT. As a result, designing a PZT-like MPB in KNN systems seems more practical. The success of constructing a tricritical triple point in BaTiO<sub>3</sub> system [74,75], as well as similar attempt in KNN-based ceramics [54], could probably offer fresh and innovative ideas for the future.

### 3 Sintering techniques

#### 3.1 Spark plasma sintering (SPS)

Sintering has been among the most active topics for KNN-based materials ever since it was first discovered half a century ago [63,76,77], due to intrinsic volatility of alkaline species. Thus, normal sintering (in contrast to pressure-assisted sintering like hot-pressing or spark plasma sintering) of KNN-based materials often has a chance to encounter with low density and poor microstructure, which definitely degrades piezoelectric performance. Due to the poor sinterability and high volatility of KNN, pressure-assisted sintering techniques have been utilized for pursuing high density. Jaeger and Egerton [76] firstly fabricated highly dense KNN ceramics using hot pressing and reported that the hot-pressed samples possessed high piezoelectric constants and Curie temperature. Recently, spark plasma sintering (SPS) has been increasingly used instead of conventional hot pressing because of its advantages of a rapid heating rate and short soaking time. Our group has done a series of systematic studies on SPS of KNN-based ceramics, with special emphasis on property enhancement [78-83].

Sodium potassium niobate ceramics with the nominal composition of K<sub>0.5</sub>Na<sub>0.5</sub>NbO<sub>3</sub> were fabricated successfully using SPS at a low temperature of 920 °C. By post-annealing treatment in air, the SPS-processed K<sub>0.5</sub>Na<sub>0.5</sub>NbO<sub>3</sub> ceramic samples showed typical ferroelectric and piezoelectric characteristics. The room temperature dielectric constants at 1, 10, 100, and 1000 kHz are 606, 584, 559, and 571, respectively, with dielectric loss values being in the range of 0.026 ~ 0.044. Although the grain sizes are small comparing with normally sintered samples (see Fig. 4), being on the order of 200 ~ 500 nm, the resultant K<sub>0.5</sub>Na<sub>0.5</sub>NbO<sub>3</sub> ceramics show a considerably high  $d_{33}$  of 148 pC/N. It is suggested that the enhancement is the result of

extrinsic contributions to the polarizability associated with submicron grain sizes [55]. Similar result was also found in fine-grained BaTiO<sub>3</sub> ceramics [84].

In addition to pure KNN samples, Li and Ta co-doped KNN ceramics were also investigated by SPS method [81]. Dense and fine-grained (Na<sub>0.535</sub>K<sub>0.485</sub>)<sub>1-x</sub>Li<sub>x</sub>(Nb<sub>0.8</sub>Ta<sub>0.2</sub>)O<sub>3</sub> ( $x=0.02\sim0.07$ ) (abbreviated as NK<sub>L<sub>x</sub></sub>NT) lead-free piezoceramics were fabricated by SPS at around 900 °C, followed by post-annealing in air. All the NK<sub>L<sub>x</sub></sub>NT ceramics showed single perovskite structures with a phase transition from an orthorhombic symmetry to a tetragonal one across a composition region of  $x=0.04\sim0.05$ . The sample with a composition of  $x=0.05$  had the maximum values of piezoelectric coefficient ( $d_{33}=243$  pC/N) and planar electromechanical coupling coefficient ( $k_p=46.1\%$ ), and other good properties such as  $Q_m=85$ ,  $\epsilon_r=1240$ , and  $\tan\delta=0.023$ . Because of the enhanced densification and refined microstructure, the SPS-processed NK<sub>L<sub>x</sub></sub>NT ceramics also had better fracture strength

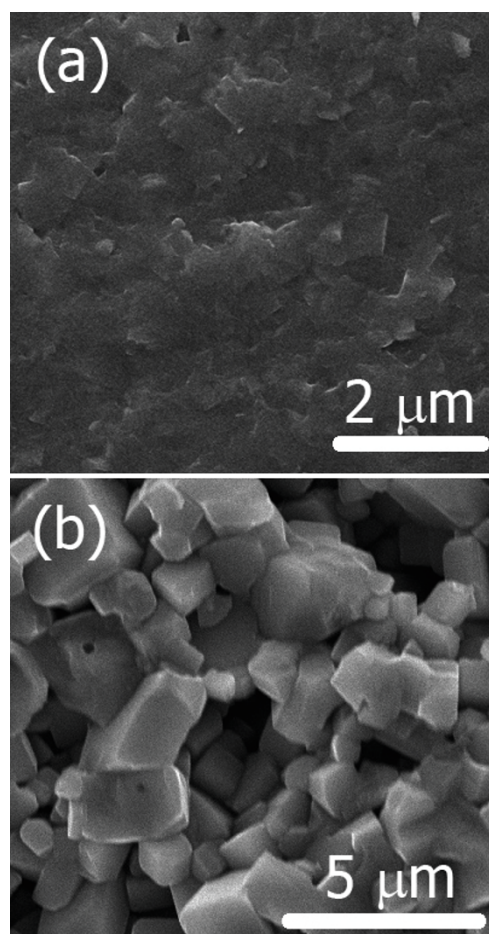


Fig. 4 SEM micrographs of (a) SPS and (b) normally sintered (K<sub>0.5</sub>Na<sub>0.5</sub>)NbO<sub>3</sub> ceramics

than that prepared by normal sintering.

### 3.2 Normal solid state sintering

Normal sintering is cost-effective and simple, which is of great attraction for mass production. So, investigation on normal sintering of KNN never ceases. It was found that in normal sintering of KNN-based ceramics, density increased greatly within a narrow temperature range but turned to decrease when the sintering temperature slightly exceeded the optimal one. Piezoelectric properties also showed similar relationship between the density and sintering temperature, but the highest piezoelectric strain coefficients were obtained at the temperatures lower than that for the highest density [85,86].

Figure 5 shows the change in the sintered densities of the KNN, Li-KNN (LKNN) and Li-Ta-KNN (LKNNT) samples as a function of sintering temperature. It was found that the change in density was very sensitive to the sintering temperature. Even though the sintering temperature was changed within a narrow range, the resultant density changed significantly for all the compositions. The density increased rapidly and then decreased slowly after reaching a maximum. Ta doping apparently shifted the maximum sintering point to higher temperature, whereas the densification behavior was almost the same both for the samples without and with Li doping. The highest density values obtained for the three compositions exceeded 95% of their theoretical densities. Fig. 5 also shows the changes in piezoelectric coefficient  $d_{33}$  of the three compositions as a function of sintering temperature. It is interesting that similar tendencies were obtained for the relationship between  $d_{33}$  and sintering temperature as the case of sintered density. All the three compositions showed their peak  $d_{33}$  values at their corresponding

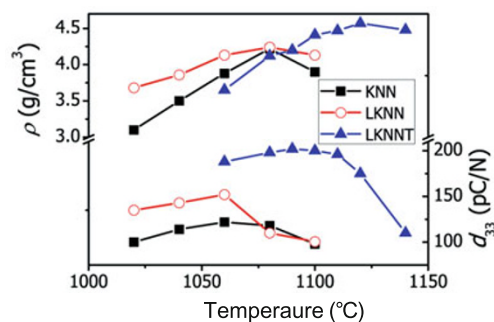


Fig. 5 Density and piezoelectric coefficient  $d_{33}$  of KNN, LKNN and LKNNT samples as a function of sintering temperature

sintering temperatures. However, it must be noted that the temperature where the maximum  $d_{33}$  value was obtained differed from the point where the peak density was achieved. For example, the highest density was obtained at 1120 °C for the LKNNT composition, whereas  $d_{33}$  of the resultant ceramics sample decreased apparently from the peak point if sintered at 1120 °C; the highest  $d_{33}$  value was obtained at the temperature which is about 20~30 °C lower than that where the highest density can be reached. In other words, the  $d_{33}$  values dropped apparently when sintered at the temperatures exceeding the points that correspond to the peak density. On the other hand, the  $d_{33}$  values of the samples were relatively high, in spite of their low density when sintered at the temperatures below the point corresponding to the peak density.

The main question about the above results is why the optimal sintering temperature for the electrical properties differed from that for the density? As shown above, the piezoelectric and dielectric properties changed with a similar tendency as the relationship between density and sintering temperature. However, the sintering temperatures where the highest  $d_{33}$  values were obtained, were not consistent with that where the highest density was reached. To answer this question, we must take into account the volatilization of alkali components during the sintering, which occurred during the sintering of KNN-based ceramics and caused the composition deviation from the starting one. The volatilization may be accelerated when the liquid phase appears, although which also enhances the densification at the same time. Our study by inductively coupled plasma (ICP) analysis confirmed the serious volatilization above 1000 °C, as shown in Fig. 6. Considering that the electric properties are sensitive to the composition change [6,79,87,88], and

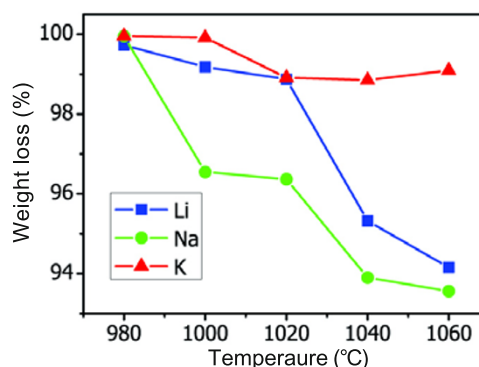


Fig. 6 Weight loss of alkaline elements in KNN-based ceramics as a function of sintering temperature

that the sample sintered to the highest density may not have the optimal composition, it is easy to understand why the highest piezoelectric and dielectric properties were obtained at somewhat lower sintering temperature than that for density. In addition, grain coarsening or abnormal grain growth occurred when the sintering temperature exceeded the peak point for the density, which may also not be favorable for the optimization of electrical properties [89].

### 3.3 Low-temperature liquid-phase sintering

Technically, the low-temperature sintering mentioned here is still in the discussion range of normal sintering. However, the densification mechanism during sintering is different due to emergency of a liquid phase, and the low-temperature sintering is greatly effective for property enhancement. As a result, we take out this topic as a separate section. As explained in the previous section, the properties of KNN-based ceramics are extremely sensitive to the processing conditions, especially the sintering temperature. The sintering temperatures are usually very high as compared with the melting points in the phase diagram, easily causing severe volatilization of alkali elements (see Fig. 6), and resulting in a composition deviation and hence property degradation. For example, in our previous study on Li-doped KNN ceramics, the  $d_{33}$  value decreases from a peak (314 pC/N) to about 220 pC/N if sintered at a temperature just 20 °C higher than the optimized one (1060 °C) for the KNN-ceramics containing 5.8 mol% LiNbO<sub>3</sub> [6]. Therefore, it is necessary to develop low-temperature sintering processes for KNN-based ceramics. Also, low-temperature sintering could satisfy the demand for co-firing KNN-based ceramics with metal electrodes for device applications. Recently, our study revealed that Li-modified KNN lead-free piezoceramics can be sintered at a temperature as low as 950 °C, whose  $d_{33}$  value is up to 280 pC/N [12,90]. This result shows that low temperature sintering and high performance could be achieved together in high-performance LiNbO<sub>3</sub>-KNN lead-free piezoceramics, which are of great potential for industrial applications. It is believed that the success of low-temperature sintering is related to the excess amount of Na<sub>2</sub>O added intentionally.

The addition of excess amount of Na<sub>2</sub>O could effectively reduce the sintering temperatures of Li-modified KNN ceramics. Figure 7 shows the shrinkage curve of  $(1-x)(\text{Na}_{0.535}\text{K}_{0.48})\text{NbO}_3$ - $x\text{LiNbO}_3$  at  $x = 0.08$ , which clearly demonstrates that the sample

started to shrink at the temperature as low as 800 °C, while the shrinkage rate had an abrupt change at 930 °C. The nearly vertical shrinkage rate curve after 930 °C suggests an unusually fast densification process, which is due to the formation of liquid phase. Although Na<sub>2</sub>O was added with excess amount, the effective sintering aid may be involved with other species. It was reported that A<sub>2</sub>O (A = Li, Na, K) could all form eutectic phases with Nb<sub>2</sub>O<sub>5</sub>, for mole ratio A:Nb more than 1:1, with melting points around 800 °C; while A<sub>2</sub>CO<sub>3</sub> (A = Li, Na, K), some of which probably remained in the matrix after calcination at 750 °C since extra Na<sub>2</sub>CO<sub>3</sub> was added initially, also has melting points lower than 950 °C. Considering that Li<sup>+</sup>, Na<sup>+</sup>, and K<sup>+</sup> could all fill into the A site of perovskite subcell, the actual liquid phase may be eutectic phases formed between any of A<sub>2</sub>O (A = Li, Na, K) and Nb<sub>2</sub>O<sub>5</sub>, or remnant carbonates A<sub>2</sub>CO<sub>3</sub> (A = Li, Na, K). The inner figures in Fig. 7 shows the SEM micrographs of  $(1-x)(\text{Na}_{0.535}\text{K}_{0.48})\text{NbO}_3$ - $x\text{LiNbO}_3$  with  $x = 0.08$  sintered at 900 °C and 950 °C, respectively. The average grain size of the sample sintered at 900 °C is about 1-2 μm, and only a few of grains are adhered together with a low density of 3.95 g/cm<sup>3</sup>. The loose microstructure indicates no appearance of liquid phase at 900 °C. By contrast, when the sample was sintered at 950 °C, the microstructure changed significantly, with some large quadrate grains of 20 to 40 μm and an enhanced density of 4.36 g/cm<sup>3</sup>. Within a merely 50 °C difference in sintering temperatures, the grain size increased from 1~2 μm to several tens of microns, confirming the effectiveness of liquid phase sintering. It has been recognized that the driving force for sintering is the decrease of surface energy of whole system. As for this case of liquid phase sintering, three

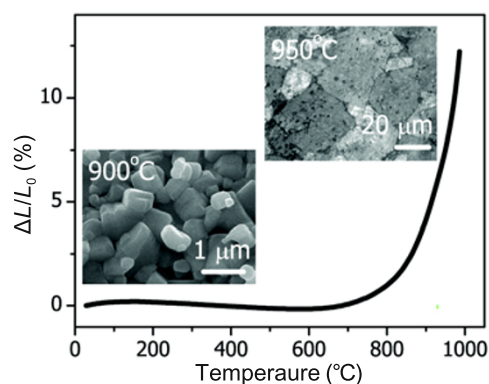


Fig. 7 Shrinkage curve and SEM micrographs of  $0.92(\text{Na}_{0.535}\text{K}_{0.48})\text{NbO}_3$ - $0.08\text{LiNbO}_3$



Table 1 Properties and phase structures of  $0.92(\text{Na}_{0.535}\text{K}_{0.48})\text{NbO}_3\text{-}0.083\text{LiNbO}_3$  samples sintered at different temperatures

Temp. (°C)	850	900	930	950	970	1000	1050	1100
$\rho$ (g/cm <sup>3</sup> )	3.40	3.96	4.27	4.38	4.32	4.16	4.13	4.06
$\rho$ , relative (%)	75.4	87.8	94.7	97.1	95.8	92.2	91.6	90.0
$d_{33}$ (pC/N)	135	195	230	280	200	125	110	100
Phase structure*	O	O-T	O-T	O-T	O-T	T	T	-

\* O is abbreviation for orthorhombic, and T is for tetragonal. “-” means ambiguous phase identification due to emergence of impurities

distinct stages have been classified [91]: (1) particle rearrangement, (2) solution-precipitation process, and (3) coalescence process. The morphology of the 950 °C-sintered sample demonstrates that it was at the final stage of liquid phase sintering, with substantially dense micro- structure and large faceted grains exhibiting shape accommodation characteristic for maximum density.

Table 1 shows the sintered density and piezoelectric coefficient  $d_{33}$  as a function of sintering temperature for the sample of  $x=0.080$ . The density increases significantly as the sintering temperature is increased from 850 to 950 °C, and reaches a peak, then decreases gradually with further increasing sintering temperature. The bulk density (about 4.38 g/cm<sup>3</sup>) obtained at 950 °C is higher than 95% of the theoretical density. The densification behavior can be also confirmed by the microstructure observation. This result is surprising because low-temperature sintering was realized by simply adding excessive Na<sub>2</sub>O, which is important for co-firing KNN-based multilayer with the Ag/Pd electrode. Although the low-temperature sintering mechanism is not yet clearly understood, liquid-phase sintering should be responsible for the densification. The grain growth behavior seems to be a clear indicator supporting our consideration about liquid-phase sintering, because the grain sizes increased exceptionally as the sintering temperature was raised from 900 to 950 °C. The density decreases when sintered above 950 °C probably because the volatilization of alkali oxides is accelerated, which is not in favor of the formation of liquid phase. In fact, the weight loss at 950 °C and 1050 °C were 0.5% and 1.3%, respectively. In addition to the highest density, a peak  $d_{33}$  was obtained when sintered at 950 °C for the  $x=0.080$  composition, also because it consisted of orthorhombic and tetragonal phases. It should be noted that for the same  $x=0.080$  composition, the resultant phase structure also changed from orthorhombic to

tetragonal with increasing sintering temperatures, as revealed in Table 1, which is the result of volatilization of alkaline species.

#### 4 Poling tricks

It is known that poling is an important process to endow ferroelectric ceramics with macroscopic piezoelectric responses, by applying a high electric field up to several thousand voltages per millimeter. Inferior ferroelectric response is obtained by insufficient poling, which is caused by various factors such as defects. The effect of poling conditions on the dielectric and piezoelectric properties of KNN-based ceramics has been a subject of recent studies [92-97], most of which pointed out that poling temperature is of significant importance for property enhancement. Nevertheless, poling process could also be utilized to manipulate interior dipole movement, which might be tuned to favorable piezoelectric performance [52,53].

For a compositionally optimized Li-doped composition, its piezoelectric coefficient  $d_{33}$  can be increased up to 324 pC/N even from a considerably high value (190 pC/N) by means of re-poling treatment after room temperature aging; such a high  $d_{33}$  value is only reachable in literature for KNN ceramics with complicated modifications using Ta and Sb dopants [46,98,99]. Table 2 shows the piezoelectric coefficient  $d_{33}$  as well as tetragonal-orthorhombic phase transition temperature ( $T_{\text{O-T}}$ ) of Li-doped KNN ceramics with respect to Li content, after both the first and second poling, respectively. Even though the sintering temperature was as low as 950 °C, a high piezoelectric constants  $d_{33}$  up to 280 pC/N was obtained at an optimal composition of  $x=0.083$  after the first poling. The basic mechanism for piezoelectric property enhancement in the present study was generally considered due to two-phase coexistence [15], which was denoted by a downward of  $T_{\text{O-T}}$  to around room



Table 2 Piezoelectric properties and tetragonal-orthorhombic phase transition temperature ( $T_{O-T}$ ) of  $(1-x)(\text{K}_{0.5}\text{Na}_{0.5})\text{NbO}_3$ - $x\text{LiNbO}_3$  after both the first and second poling

$x$	5.8	6.5	7	7.5	7.8	8	8.3	8.5
1st poled $d_{33}$ (pC/N)	125	130	145	170	185	190	280	180
2nd poled $d_{33}$ (pC/N)	128	135	170	230	260	324	255	145
$T_{O-T}$ ( $^{\circ}\text{C}$ )	155	115	105	90	68	60	43	27

temperature, as also listed in Table 2. In spite of the considerably high piezoelectric performance of Li-doped KNN ceramics after the first poling treatment, the improvement of  $d_{33}$  after the second poling process is much more attractive, with a peak  $d_{33}$  of 324 pC/N at  $x=0.080$ , as shown in Table 2. It should be noted that more and more increment was made in  $d_{33}$  after the second poling with increasing Li content, with a maximum of 134 pC/N (from 190 to 324 pC/N) at  $x=0.080$ . The maximum increment rate was amazingly more than 70%, even though no special treatment was given to the samples but just re-poling after aging for two months. However, the sample with an excellent  $d_{33}=280$  pC/N at  $x=0.083$  after the first poling showed no increase in  $d_{33}$  after re-poling two months later, but its  $d_{33}$  turned to decrease by 25 pC/N. It seems that aging and re-poling could not increase  $d_{33}$  for the  $x \geq 0.083$  samples. In view of the  $T_{O-T}$  and XRD measurements, it could be easily concluded that samples with coexisting phases at room temperature are favorable for the aging and re-poling induced enhancement of piezoelectricity, while those dominated by orthorhombic (with  $T_{O-T}$  more than  $120^{\circ}\text{C}$ ) or tetragonal phases are neither sensitive nor profited. For the ferroelectric properties, it was found that the second poling resulted in an increasing  $P_r$  from 6.78 to  $14.46 \mu\text{C}/\text{cm}^2$ , and a reduced  $E_c$  from 22.52 to  $17.58 \text{ kV}/\text{cm}$ , respectively. The striking increase of  $d_{33}$  in the present study was due to large improvement of  $P_r$  [100].

As mentioned above, the increase of  $d_{33}$  was due to large improvement of  $P_r$ , which was accomplished by domain switching. Domain switching does not change the crystalline structure of the ceramics, but do make striking influences on macroscopic orientations of local atoms arrangement. XRD technique can provide accurate information concerning these variations by relative peaks intensity change, then domain evolution processes can be deduced. Figure 8 shows the XRD patterns of (222) and (004) lattice plane series for the  $x=0.080$  sample, which were taken both before and

after the first poling, as well as after the second poling, respectively. The peak indexing is adopted for a tetragonal phase. These high-angle X-ray diffractions offer phase information more precisely than the low-angle ones usually employed in literatures [67].

As shown in Fig. 8, for the as-sintered sample, higher intensity of the (400)/(040) peak over (004) peak is a clear symbol for the tetragonal phase since  $c > a = b$ ; however, slight split of the (222) peak, and over-lifted  $k\alpha_2$  line of (004) peak indicate the presence of the orthorhombic phase ( $a = c > b$  with  $90^{\circ} < \beta < 91^{\circ}$ ). After the first poling treatment, intensity contrast between (400)/(040) peak and (004) peak reverses, which is easily understood for the tetragonal phase due to an increase of c-domains. Besides, intensity increment of (004) peak after poling is also observed for the orthorhombic phase, despite the fact that  $\langle 001 \rangle$  is not the spontaneous polarization direction. The increased intensity of (004) peak in orthorhombic phase can be easily deduced as a collateral consequence of domain rotation towards  $\langle 110 \rangle$  directions. It should be emphasized that, significant

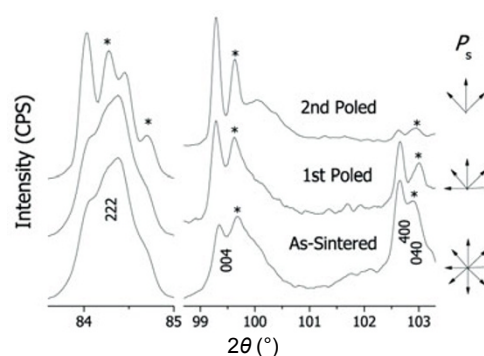


Fig. 8 Patterns of (222) and (004) lattice plane series for the sample  $0.92(\text{Na}_{0.535}\text{K}_{0.48})\text{NbO}_3$ - $0.08\text{LiNbO}_3$ , both before and after the first poling, as well as after the second poling, respectively. The  $P_s$  evolution inside domains is listed aside, assuming that the electric field during poling is parallel to  $\langle 001 \rangle$  direction

changes, represented by distinct split of (222) peak and almost disappearance of (400)/(040) peak, took place after the second poling. The deduced  $P_s$  evolution inside domains is listed aside in Fig. 8, assuming that the electric field during poling is parallel to the  $\langle 001 \rangle$  direction. The  $P_s$  parallel to  $\langle 100 \rangle$  and  $\langle 010 \rangle$  directions after the first poling, switched to  $\langle 001 \rangle$  direction after the second poling, which indicates that the non-switched  $90^\circ$  domains successfully rotated and kept constant by means of the aging and re-poling process. A mechanism corresponding to details of spontaneous polarization change in the domain level was proposed, concerning the combined effect of migration of oxygen vacancies, and interaction between defect dipoles and spontaneous polarization inside domains, which can be found elsewhere in detail [52].

It is widely accepted that poling of KNN-based ceramics is usually accompanied with problems, such as large leakage current, easy breakdown, etc.; however, the above results demonstrate that poling is also a chance which could be resorted to for property enhancement. If poling condition could be optimized for a specific composition, very high  $d_{33}$  can be obtained without complicated doping compositions.

## 5 Summary

KNN-based ceramics have been investigated as a potential candidate for a new group of lead-free piezoelectric materials, and recent advances related to phase transitions, sintering techniques and property enhancement are discussed. Despite inferior piezoelectric response than PZT, KNN system is still favored by high Curie temperature, complete environment-friendliness, etc. The most urgent issue concerning development of KNN-based materials should be a breakthrough, both theoretically reasonable and experimentally testified, in property enhancement in the absence of PPT effect, such as real PZT-like MPB design. Then problems like temperature instability could be solved thoroughly. Also, secondary considerations, such as poling conditions, processing optimization, domain influences, etc., would be favored as well, due to request both practically and scientifically. Moreover, from the viewpoint of industrial application,  $d_{33}$  is not the only figure of merit. Actually, preliminary reports on piezoelectric devices utilizing KNN-based materials already exhibit

promising results, including high-frequency ultrasonic transducers [101-103], surface acoustic wave (SAW) devices [29] and so on [104-107]. Although it is possible that KNN cannot carry out the historical assignment to substitute PZT everywhere, it must find suitable opportunities to fulfill itself eventually.

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